



## **Experimental and Numerical Studies of Aquathermolysis Tests of Christina Lake Bitumen with Steam**

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### **Summary**

H<sub>2</sub>S emissions from thermal recovery operations of bitumen and heavy oil are becoming an increasing concern. Many SAGD programs in northern Alberta are finding H<sub>2</sub>S level rising as well aging. The generation of H<sub>2</sub>S is due to an aquathermolysis reaction of bitumen or heavy oil with high temperature steam that converts sulfur-containing compounds in bitumen, minerals (pyrite), sulphate into H<sub>2</sub>S. The amount of H<sub>2</sub>S depends on forms of sulphur, types of reaction, kinetics, and may vary from reservoir to reservoir.

This study used Christina Lake bitumen and experimentally determined the H<sub>2</sub>S onset temperature upon reacting with steam for 3 days. Then the amount of H<sub>2</sub>S and other gases generated by aquathermolysis, including CO<sub>2</sub>, CO, H<sub>2</sub>, and light hydrocarbons were quantified and detailed analysis of the post-reaction oil composition were performed. The temperature effect was studied on the amount of generated gases at 212, 223, and 242°C, and reaction time effect for 3, 30 and 45 days respectively. The results show the amount of generated gas increases with both reaction temperature and reaction time. The experimental data are used to create a kinetic predictive model which can be further implemented into the compositional and thermal simulation to predict components' variation in the field scale. The bitumen after aquathermolysis were measured for its viscosity at "live" conditions, and it was found that the viscosity slightly decreased with reaction time at the same reaction temperature.

### **Acknowledgements**

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### **References**

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